RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College affiliated to University of Calcutta)

B.A./B.Sc. SIXTH SEMESTER EXAMINATION, MAY 2017 THIRD YEAR [BATCH 2014-17] INDUSTRIAL CHEMISTRY (Honours)

Date : 02/05/2017 Time : 11 am - 1 pm

Paper: VII (U I & IV)

[Use a separate Answer Book for each Unit]

Full Marks: 50

<u>UNIT-I</u>

	<u> </u>			
1.	An	nswer <u>any eight</u> questions: Discuss the current Indian Heavy Chemical Industries scenario and its future prospects.	[5×8] 5	
2.	a)b)c)	Write the chemical reactions involved in the contact process of manufacture of sulphuric acid. Draw the flow diagram of the contact process. Explain why SO ₃ is absorbed in 98% H ₂ SO ₄ not in water?	2+2+1	
3.	a)b)c)	Write the physico-chemical principles of the nitric acid manufacturing process. Discuss the nitric acid manufacturing process in details. Write the use of nitric acid.	2+2+1	
4.		Describe wet process of phosphoric acid manufacture. Write all the chemical reactions involved. How Uranium is recovered from phosphate rock?	3+1+1	
5.	b)	Name the different types of electrolytic cells which are used for the production of caustic soda and chlorine. Why mercury cells are not used in Industry now-a-days? Draw a neat flow diagram of membrane cell process for caustic soda and chlorine manufacturing process. Write the name of the membrane used.	2+2+1	
6.		Describe the manufacturing process of Soda Ash by modified Solvay process with respect to a) Raw material used; b) Chemical reactions involved; c) Flow diagram.	5	
7.		Explain briefly important engineering process parameters involved in a typical urea manufacturing plant starting from Naphtha / Natural gas. Name three large manufacturing units in India and their location.	5	
8.	a) b)	What is meant by the term N-P-K used in labelling the fertilizers. Write in detail the manufacturing process of Triple super phosphate.	2+3	
9.		Choose the correct answer: a) Calcium Carbide manufacturing process require the temperature i) 300 – 500°C ii) 750 – 900 °C iii) 2000 2200 °C	+1+1+1	

iv) 3500 - 4000 °C

b) Silicon carbide is an abrasive material ii) is an adhesive material iii) is a glass iv) is a ceramic c) For nitrogenous fertilizer the fertilizer value is expressed as i) Nitrogen content ii) Potassium content iii) KNO₃ content iv) HNO₃ content d) Ostwald process is used to manufacture of i) H₂SO₄ ii) H₂PO₄ iii) HNO₃ iv) CaC e) Dry chlorine can be handled in a vessel made of i) PVC ii) Steel iii) Brass iv) Nickel 10. Write down the method of manufacture of the following (any two) 2.5+2.5a) Graphite b) Bleaching powder c) Chloramine-T d) Sodium thiosulphate e) Silicon carbide **UNIT-IV** Answer **any two** questions: $[5\times2]$ 11. a) Why lacquers dry faster than oleo resinous varnish? b) Name pair of pigments for following colours: (i) Black (ii) Blue (iii) Green (iv) Red c) What is 'Paint Drier'? Give an example. d) Name a Paint Thinner. 1+2+1+112. a) What do you mean by the term: Pigment volume concentration (PVC)? b) 320 lit. of a paint was analysed as under: Pigment and extenders (Litres) $TiO_2(R)$ 10 Blane Fix 3 Phthalocyamnine Blue 3 Carbon Black 1 Precipited chalk 3 Vehicle (Binder)

Butylated melamine -

Formaldehyde resin in Xylol 100 (50% solid) **Solvents** A mixture of Tolvol, mixed 200 xylene and C-9 solvent Total 320 lit. Evaluate PVC and opine on the nature of paint. 1+413. a) Define the term: Insecticide and Herbicides. b) Name one each of Inorganic, Natural and Synthetic insecticides. c) Outline the synthesis of Benzene Hexachloride (Gammaxace, ICI). 1+2+214. Write short notes on (any two): 21/2+21/2 a) Plastic Emulsion paints b) Water Thinnable Emulsion paints c) Varnish (oleo resinous and spirit both) d) Anticarrosive Bituminous coating (Black Japan) ____×___